GLOSSARY AND CREDITS

Our world is in a constant state of decaying and regenerating itself. As a chemistry student you have a great opportunity to see and study this principle at work. Equilibrium systems exist throughout nature. You are a walking equilibrium system. When your system is out of equilibrium, you are probably sick. The force of increasing entropy countering the force of decreasing enthalpy is at the heart of our study of equilibrium systems.

Internet Links

Click here for more on Calculations with Moles and Mass

Phase Changes

Click here for more on Solutions and Mixtures

Electrochemistry

Click here for more information on Solubility and Equilibrium

Acids and Bases and pH Practice Problems

Acids and Bases